# An X-ray Investigation of the High-Temperature Phase of K<sub>2</sub>SnCl<sub>6</sub>

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### Abstract

At room temperature the background of X-ray singlecrystal photographs exhibits diffuse streaks along the line  $\Delta$  from  $\Gamma$  to X and less sharp diffuse domains in the directions  $\Sigma$ . The streaks are explained by a soft rotary phonon  $A_2(\Delta)$  corresponding to an in-phase rotation of the octahedra in a plane. The displacements of the Cl ions due to their librations are identified in a difference-Fourier chart. The second diffuse system is explained by a transversal accoustical mode  $B_1(\Sigma)$ . The in-phase rotation of the octahedra in a plane minimizes the electrostatic energy, independently of the phase of the rotation in adjacent planes. In the model the repulsive parameter between the Cl and K ions decides whether the space group P4/mnc, as a result of an  $A_{2e}(X)$  condensation, or the orthorhombic Pnnn, as a result of a second rotary mode condensation, is energetically favorable.

## 1. Introduction

Crystals of the chemical compositon  $R_2MX_6$  having the cubic antifluorite structure (Fig. 1) may be thought of as being composed of 'isolated'  $MX_6$  octahedra which are centered in the (000) position of a f.c.c. lattice. The R ions at  $(\frac{1}{4}, \frac{1}{4}, \frac{1}{4})$  are surrounded by  $MX_6$  octahedra in a tetrahedral coordination. For K<sub>2</sub>SnCl<sub>6</sub>, the compound we are concerned with here, all the structural information necessary is contained in Table 1. Two parameters are sufficient to describe the geometrical aspects of the structure: the lattice constant  $a_0$  and the position of the chlorine ion, (u00). From the numerical value u = 0.243 (this work) it is immediately clear that the idealization of isolated octahedra is not a very realistic one; the chlorine ions are only slightly closer to Sn (0.243 Å) than to the octahedral holes (0.257 Å) at  $(\frac{1}{2}00).$ 

The structure described here is closely related to the perovskite (e.g. SrTiO<sub>3</sub>) and the elpasolite (e.g.  $K_2NaAlF_6$ ) structures. In the first case all the octahedral positions at (000) and ( $\frac{1}{2}$ 00) are occupied by the same ions (e.g. Ti), and  $u = \frac{1}{4}$  by symmetry. It is of

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course more appropriate to describe this structure as a P lattice with  $a = \frac{1}{2}a_0$  (Fig. 1, Table 1). In the elpasolites the two octahedral positions are occupied by different ions (e.g. Na, Al).

All the structures mentioned above are unstable with respect to displacive phase transitions which have been investigated mostly for the perovskites. In the hightemperature phase of  $K_2 SnCl_6$ , a librational model of the  $MX_6$  octahedra was deduced from X-ray photographs and neutron experiments (Boysen, Ihringer, Prandl & Yelon, 1976). It is suggested that the librational motion of the  $MX_6$  octahedra plays an important role in the phase transitions of such compounds: Morfee, Staveley, Walters & Wigley (1960) and other authors (Sasane, Nakamura & Kubo, 1970; Jeffrey, 1972; Boysen, Ihringer, Prandl & Yelon, 1976: Winter, Rössler, Bolz & Pelzl, 1976; Boysen,



Fig. 1.  $K_2$ PtCl<sub>6</sub> structure (from D'Ans-Lax, 1970): small spheres, K; octahedra, PtCl<sub>6</sub>.

## Table 1. Positional and thermal parameters for K<sub>2</sub>SnCl<sub>6</sub> at 300 K

Lattice constant a = 9.988 Å

8 K in (c)	$(0.25, 0.25, 0.25) u_{11} = 0.054 (1)$	
4 Sn in (a)	$(0, 0, 0) u_{11} = 0.0206 (3)$	
24 Cl in (e)	$[0.2407(2), 0, 0] u_{11} = 0.021(1)$	$u_{22} = 0.082$ (2)

The thermal parameters in Å<sup>2</sup> are the mean-square displacements in the directions of the corresponding axes. To get the  $B_{ik}$ parameter, multiply by  $8\pi^2$ .

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1977; Ihringer, 1977) explain the transition in  $K_2SnCl_6$ at  $T_{c1} = 261$  K by the softening of a rotary lattice mode. A direct observation of this mode in the cubic phase of  $K_2SnCl_6$  by inelastic neutron scattering is reported by Boysen (1977) and Vogt (1979). The lattice dynamics of  $R_2MX_6$  compounds were studied for the following systems:  $K_2ReCl_6$  by O'Leary & Wheeler (1970), also more recently by Lynn, Patterson, Shirane & Wheeler (1978);  $Cs_2UBr_6$  by Chodos & Satten (1975); and  $Cs_2SiF_6$  by Patterson & Lynn (1979).

In perovskites many of the displacive phase transitions from the cubic to the tetragonal or orthorhombic phase are associated with rotations of the  $MX_6$  octahedra. Dénoyer, Comès, Lambert & Guinier (1974) and Dénoyer (1977) observed a streak system caused by a rotational mode on X-ray photographs of NaNbO<sub>3</sub>.

We will see that the relationship between the  $R_2MX_6$ and the perovskite structures yields closely related atomic displacement fields for the energetically lowest rotational modes of the octahedra. However, besides this common feature, each structure shows a second characteristic diffuse system in X-ray photographs. In the perovskites  $RMX_3$  there are diffuse planes caused by the one-dimensional M chains. In K<sub>2</sub>SnCl<sub>6</sub> we find diffuse scattering along the  $\Sigma$  direction ( $\Gamma$ -K-X; c.f. Fig. 4, Fig. 5) which can be explained by a low-lying transversal acoustical mode.

#### 2. High-temperature structure

#### 2.1. Experimental

Diffraction data were measured with filtered Mo  $K\alpha$  radiation on a Siemens three-circle diffractometer operated in the  $\theta/2\theta$  scan mode  $(\sin \theta/\lambda \le 1.2 \text{ Å}^{-1})$ . Two data sets (referred to as DS1 and DS2 in the following) of 1200 reflections each were taken from two different octahedral crystals both with 0.5 mm diameter. For the numerical absorption correction, the distances between the boundary crystal faces were measured optically.

### 2.2. Data reduction and structure refinements

The absorption correction, data reduction and structure refinement was performed with the program SHELX 76 (Sheldrick, 1975). Equivalent reflections in both data sets were merged to give 264 and 266 independent reflections.

Scattering curves for the Cl<sup>-</sup>, K<sup>+</sup> and Sn<sup>4+</sup> ions were taken from *International Tables for X-ray Crystallog-raphy* (1974). Refinements with the final residuals R = 0.047,  $R_w = 0.035$  for DS1 and R = 0.039,  $R_w = 0.033$  for DS2 were computed with the full data sets

and weights from counting statistics. The final parameters of DS1 and DS2 agree within their standard deviations (e.s.d.), the Cl x/a and the Sn thermal parameter within  $2 \times \text{e.s.d.}$  The parameters from the second data set are listed in Table 1.

The results agree well with those derived by Lerbscher & Trotter (1976) from X-ray and Boysen & Hewat (1978) from neutron powder data. The bondlength correction (Cruickshank, 1956) for the librating Cl ions gives  $5 \cdot 8^{\circ}$  as the amplitude of the rotation angle, the corrected Cl parameter is 0.2434 (4).

In DS2 there are four reflections with  $F_o - F_c$  greater than  $2\sigma$ : 111, 002, 004, 044. The 111 and 002 reflections are assumed to be of poor quality because they are inconsistent with those of DS1, the strongest reflections, 004 and 044, in both DS1 and DS2 are weakened by extinction. Omitting these four reflections improved the residuals to R = 0.028,  $R_w = 0.025$ , and refinement shifted the parameters of Table 1 by less than  $2 \times \text{e.s.d.}$ in the direction of the results obtained with the first data set. The Sn thermal parameter was shifted about  $6 \times$ e.s.d. The final R was 0.022, the refined weighting scheme of  $3.5/\sigma^2$  gave  $R_w = 0.019$ .

## 2.3. Difference electron densities

The difference-Fourier chart of the plane (0yz) in Fig. 2 was computed with the second data set (DS2) with the four reflections 111, 002, 004, 044 omitted.\* Fourier charts usually suffer from series termination errors. In order to test their influence on the difference charge shown in Fig. 2, the computation was repeated using reflections only up to  $\sin \theta / \lambda = 0.90$ . The resulting map shows the same details as Fig. 2 computed from the complete data set, the only difference being that the extrema are less pronounced. They represent, therefore, true information on the charge distribution. The density modifications along the principal axes of the Cl and Sn ions indicate that the anisotropic temperature factors in the harmonic approximation are not sufficient to describe the displacements. The density maxima near the Cl and Sn ions parallel to the bond Sn-Cl give evidence for an acoustic phonon of low energy. Note that the maxima of the Cl ions perpendicular to the bond have no analog near the Sn ions. Therefore, the arc-shaped maxima around the Sn ions may be produced by the librating octahedra. Another reason for these maxima could be modification of the electronic density by covalent bonds between the ions of the octahedra. In the light of

<sup>\*</sup> Lists of the structure factors for the three refinements, the list of resulting parameters and the difference-Fourier chart computed with the full data set DS2 have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 34667 (9 pp.). Copies may be obtained through The Executive Secretary, International Union of Cystallography, 5 Abbey Square, Chester CH1 2HU, England.

inelastic neutron measurements (Boysen, 1977; Vogt, 1979) which verify the soft librational mode along the direction  $\Delta$  (see Fig. 5), I favor the first explanation.

3. Diffuse background in X-ray single-crystal photographs

X-ray single-crystal photographs of  $K_2SnCl_6$  exhibit two systems of strong, diffuse scattering. In Fig. 3 the first system shows small streaks parallel to the principal



Fig. 2. Difference-Fourier density map in the (001) plane of  $K_2SnCl_6$ . The dotted regions are positive areas, broken lines denote negative contours. The density contour lines are separated by 30 units, where 160 units =  $1 e/\dot{A}^3$ .



Fig. 3. Precession photograph from  $K_2SnCl_6$  at room temperature. First layer of the [001] zone with filtered Mo  $K\alpha$  radiation. The  $a^*$  axis lies horizontal.



80

axes along the direction  $\Delta(\Gamma \rightarrow X)$  (see Fig. 4, Fig. 5).





(b)

156

along the streaks is small. In real space we therefore expect correlations in planes only. The small diameter of the streaks indicates a long extension of the correlations within the planes.

The second system, pointing along the  $\Sigma$  direction  $(\Gamma - K - X)$  (Fig. 4), is broader than the first one. Its intensity is proportional to the intensity of the corresponding Bragg reflections and therefore it must be due to acoustic phonons with neighboring atoms moving in phase. Fig. 5 shows the diffuse scattering with respect to the Brillouin zones.

To interpret both systems by low-lying phonons, the basis vectors of the normal-mode displacement fields were evaluated for the direction  $\Delta$  (Figs. 5, 6a) and  $\Sigma$ (Fig. 6b) by the projections operator technique (Maradudin & Vosko, 1968). Table 2(a) and (b) shows the eigenvector-dependent part of the inelastic scattering factors of the Cl ions for all normal modes. Rigid octahedra are assumed in these calculations. The  $A_{2}$ mode completely describes the observed extinction rules and the intensity of the streak system in the  $\Delta$ direction. The second system in the vicinity of evenindexed reflections shows maximal intensity along [110] for *hhl* reflections with h = 4n, and zero intensity along the same direction for hhl reflections. The transverse acoustic  $B_1$  mode along  $\Sigma$  explains these diffuse intensities. At the X point, the  $A_{2g}$  mode is compatible with the  $A_2(\Delta)$  mode as well as with a  $B_1(\Sigma)$ mode.



Fig. 5.  $\Sigma$  and  $\Delta$  directions in neighboring Brillouin zones for the face-centered cubic lattice in the vicinity of even- and odd-indexed reflections.

In Fig. 7 a cross section through the Cl<sub>6</sub> octahedra is shown for three typical situations: for all octahedra in their time-averaged position ( $\varphi = 0$ ); for an in-phase rotation and for an anti-phase rotation of nearest octahedra. The second and third cases correspond to  $A_2(\Delta)$  and  $E(\Delta)$  modes, respectively. For the  $A_2(\Delta)$  mode there is only little steric hindrance up to a rotation of 14° (for the Cl ionic radius chosen), for the  $E(\Delta)$  mode, however, repulsion occurs even at a small rotation angle. From this one can conclude that in an  $A_2(\Delta)$  mode displacement there are only weak forces between the Cl ions belonging to neighbouring octahedra, therefore the mode should be very soft.

Octahedra of different planes are coupled by the K ions between them. The shortest K-Cl distance is 3.54 Å but the sum of the ionic radii is only 3.14 Å. Therefore, the octahedra can rotate around [001] by about  $14^{\circ}$  before K and Cl come into direct contact. This may explain the weak coupling of adjacent planes for the  $A_2$ -mode displacement field.

To get more quantitative information the latticeenergy calculations of Morfee, Staveley, Walters & Wigley (1960) were redone and generalized beyond a strictly electrostatic approach by adding a Born–Meyer potential term

$$\Phi_{ab} = -q_a q_b / r_{ab} + \lambda_{ab} \exp\left(-r_{ab} / \rho_{ab}\right).$$

 $\Phi_{ab}$  denotes the potential energy of two particles a and b with charges  $q_a$ ,  $q_b$  at a distance  $r_{ab}$ .  $\lambda_{ab}$  and  $\rho_{ab}$  are the repulsive parameters. The lattice sums of the Coulombic part of the energy were evaluated by Ewalds method (Ewald, 1921; Leibfried, 1955), the repulsive parts were summed in real space. Two sets of calculations were done: the first with the octahedra in the crystal rotating in the same sense around an axis parallel to b corresponding to the  $A_{2g}(\Gamma)$  mode and a second with the rotation in adjacent planes out of phase by 180°  $[A_{2e}(X) \mod e]$ . The parameter of greatest influence on the potential turned out to be the repulsive constant  $\rho_{ab}$  between the K and Cl ions. Starting with the parameter reported for KCl crystals (Born & Huang, 1954),  $\rho_{\rm KCl} = 0.324$ , the potential shows a deep minimum at  $\gamma = 20^{\circ}$  [Fig. 8(1)] which agrees with Morfee's result (Morfee, Staveley, Walters & Wigley, 1960), but this minimum vanishes if  $\rho_{KCI}$  is raised above about 0.36 [Fig. 8(2)]. A similar change in the Cl-Cl repulsive parameter of neighboring octahedra hardly affects the depth of the minimum for the  $A_{2p}(X)$  displacement field. This agrees with O'Leary & Wheeler (1970), who stated in their calculations of the dynamics that a change in the force constants between neighboring octahedra strongly affects the longitudinal rotary mode  $E_g$ , but leaves the transversal mode  $A_{2g}$ unaffected.

There is no significant difference in energy between an in-phase rotation of all octahedra  $[A_{2\ell}(\Gamma)]$  and an alternating rotation in adjacent planes  $[A_{2g}(X)]$ . This indicates that neighboring planes are nearly uncoupled for this type of motion. We therefore expect a flat dispersion curve for the  $A_2(\Delta)$  mode, which is in agreement with the X-ray observation described earlier [Fig. 8(3)]. No minimum in the energy appears if adjacent octahedra in a plane are tilted in opposition to one another, according to the  $E_g$  mode at the X point.

In this simple model the numerical value of  $\rho$  is crucial for deciding which structure is established after cooling the crystal: the tetragonal structure with P4/mnc, resulting from the  $A_{2g}(X)$  condensation; or an orthorhombic structure with four molecules in the unit cell (Kugler, 1979). Such a structure could be explained by two successive phase transitions: first an  $A_{2g}(X)$ condensation with wave vector (001), followed by a second rotation of the octahedra with wave vector (010) around an axis [010] normal to that of the first rotation. The symmetry P4/mnc after the first transition would be reduced by the second to *Pnnn*. However, this model cannot explain details of the transitions, e.g. that at  $T_{c1} = 261$  K we find an  $A_{2g}(X)$  and not an  $A_{2g}(\Gamma)$  condensation; this could be, for example, because all ions are treated as point charges and their polarizability remains unconsidered.

### 4. Discussion

We have found that in  $K_2 SnCl_6$  crystals the 'in-phase' rotation of neighboring octahedra in (001)  $\bigcirc$  planes minimizes the lattice energy. From the narrow streaks, on the other hand, long-range correlations of the rotation in these planes is deduced, whereas the coupling between adjacent planes is weak. There is indeed a small smooth increase of the streak intensity around the X point which indicates a slight decrease of the mode energy  $\omega(q)$  at X (Ihringer, 1977), that is, the antiphase rotation of adjacent planes is preferred.



Fig. 6. Normal-mode displacement fields for the lattice complexes in  $K_2SnCl_6$ . The basic translation vectors are  $(\frac{11}{20})$ ,  $(0\frac{11}{22})$  and  $(\frac{1}{2}0\frac{1}{2})$ , so O and  $\bullet$  denote translation-equivalent K ions. A translation along r yields a phase shift of  $\cos 2\pi k \cdot r$ ; k is the wave vector. (a) k lies in the direction  $\Delta$  parallel to the b axis, its point group is 4mm. The K<sup>+</sup> displacements with a phase factor -1 between equivalent K<sup>+</sup> ions are shown for k near the X point. For the two-dimensional representation E of each displacement field, only one basis vector is shown, the second can be derived by exchanging the a and c axes. (b) k lies in the  $\Sigma$  direction ( $\Gamma - K - X$ ; Fig. 5). The K<sup>+</sup> displacements are shown for  $\mathbf{k} = (\frac{11}{2}0)$ .

# Table 2. Mode inelastic structure factor $\hat{F}$ for the rigid $SnCl_6$ octahedra of $K_2SnCl_6$ as a function of the wave vector $\boldsymbol{q}$ for different normal modes

 $\hat{F} = \sum [\mathbf{k}, \mathbf{e}_j(\lambda \mathbf{q})] e^{i\mathbf{k}\mathbf{r}j}$ , where  $\mathbf{e}_j(\lambda \mathbf{q})$  is the eigenvector of the *j*th atom for phonon branch  $\lambda$  of the wavevector  $\mathbf{q}$ ,  $\mathbf{k} = \mathbf{H} + \mathbf{q}$ ,  $\mathbf{H}$  is the Bragg point vector and  $\mathbf{r}_j$  is the position of the *j*th atom.

(a) Wave vectors parallel to [010] along the direction  $\Delta$  of the Brillouin zone,  $\delta$  is the deviation of the anion parameter from 1/4

⊿ <b>k</b> ∥ {010}		Acoustical modes	Optical modes		
h	k	l	$ \begin{array}{ccc} \overline{A_1(\varDelta)} & E^{(2)}(\varDelta) \\ \overline{(k+q)} \times & h \times \end{array} $	$A_2(\Delta)$	$E^{(1)}(\varDelta)$
4 <i>n</i>	4 <i>n</i>	4 <i>n</i>	$2 + \cos(\pi/2) q$	0	$-l\cos(\pi/2)q$
4n + 2	4 <i>n</i>	4 <i>n</i>	$\cos(\pi/2)q$	$2\pi h l\delta$	$-l\cos(\pi/2)q$
4n + 2	4 <i>n</i>	4n + 2	$-2 + \cos(\pi/2) q$	0	$-l\cos(\pi/2)q$
4 <i>n</i>	4n + 2	4 <i>n</i>	$-2 + \cos(\pi/2) q$	0	$l\cos(\pi/2)q$
4n + 2	$4n \cdot 2$	4 <i>n</i>	$-\cos(\pi/2)q$	$2\pi h l\delta$	$l\cos(\pi/2)q$
4n + 2	4n + 2	4n + 2	$-2 - \cos(\pi/2) q$	0	$l\cos(\pi/2)q$
4n + 1	4n + 1	4n + 1	$-\sin(\pi/2)q$	l-h	$-k-q+l\sin(\pi/2)q$
4n — 1	4n + 1	4n + 1	$-\sin(\pi/2)q$	l + h	$-k-q+l\sin(\pi/2)q$
4n - 1	4n + 1	4n — 1	$-\sin(\pi/2)q$	l-h	$-k-q+l\sin(\pi/2)q$
4n + 1	4n - 1	4n + 1	$\sin(\pi/2)q$	l-h	$-k-q-l\sin(\pi/2)q$
4n - 1	4n - 1	4n + 1	$\sin(\pi/2)q$	l + h	$-k-q-l\sin(\pi/2)q$
4 <i>n</i> – 1	4n - 1	4n - 1	$\sin(\pi/2)q$	l-h	$-k-q-l\sin(\pi/2)q$

(b) Wavevectors q parallel to [110] along the direction  $\Sigma$  of the Brillouin zone.

<b>Σk</b> ∥[110]			Acoustical modes	Optical modes	
h	k	l	$ \frac{A_1(\Sigma)}{(h+k+2q)\times} \begin{array}{cc} B_1(\Sigma) & B_2(\Sigma) \\ (h-k)\times & l \times \end{array} $	$\overline{A_2(\Sigma)}$	$E(\Sigma)$
$   \begin{array}{r}     4n \\     4n + 2 \\     4n + 2 \\     4n \\     4n + 2 \\     4n + 2 \\     4n + 2 \\     4n + 2 \\   \end{array} $	4n 4n 4n + 2 4n 4n 4n + 2	4n 4n 4n 4n + 2 4n + 2 4n + 2 4n + 2	$1 + 2 \cos (\pi/2) q$ $1$ $1 - 2 \cos (\pi/2) q$ $-[1 - 2 \cos (\pi/2) q]$ $-[1 + 2 \cos (\pi/2) q]$	$0$ $l \sin (\pi/2) q$ $0$ $0$ $-l \sin (\pi/2) q$ $0$	$(h-k)\sin(\pi/2)q$ $(h+k)\sin(\pi/2)q$ $(h-k)\sin(\pi/2)q$ $(h-k)\sin(\pi/2)q$ $(h+k)\sin(\pi/2)q$ $(h-k)\sin(\pi/2)q$
$ \begin{array}{r} 4n + 1 \\ 4n - 1 \\ 4n - 1 \\ 4n + 1 \\ 4n - 1 \\ 4n - 1 \\ 4n - 1 \end{array} $	$ \begin{array}{r} 4n+1\\ 4n+1\\ 4n-1\\ 4n+1\\ 4n+1\\ 4n-1\\ \end{array} $	$ \begin{array}{r} 4n+1\\ 4n+1\\ 4n+1\\ 4n-1\\ 4n-1\\ 4n-1\\ 4n-1\\ 4n-1\\ \end{array} $	$\begin{array}{c} -2 \sin (\pi/2) q \\ 0 \\ 2 \sin (\pi/2) q \\ -2 \sin (\pi/2) q \\ 0 \\ 2 \sin (\pi/2) q \end{array}$		$ \begin{array}{l} (h-k)\cos{(\pi/2)} q \\ (h+k)\cos{(\pi/2)} q \\ (h-k)\cos{(\pi/2)} q \\ (h-k)\cos{(\pi/2)} q \\ (h-k)\cos{(\pi/2)} q \\ (h+k)\cos{(\pi/2)} q \\ (h-k)\cos{(\pi/2)} q \end{array} $

In the perovskites on the other hand there are planes perpendicular to [100] with rigid octahedra in antiphase rotation that are linked by a common anion. It is a particular feature of the rotational modes that they induce nearly identical anion-displacement fields in both structures. The in-phase rotation of neighboring



Fig. 7. Displacement of the Cl ions in the (001) plane. The ionic radii ( $r_{\rm Cl} = 1.81$  Å) are scaled with the unit cell (a = 10 Å). (a) Anti-phase rotation of adjacent octahedra. (b) Octahedra at rest with  $\varphi = 0$ . (c) In-plane rotation of adjacent octahedra; neighboring octahedra may rotate with little hindrance up to  $\varphi \simeq 15^{\circ}$ .

SnCl<sub>6</sub> groups in K<sub>2</sub>SnCl<sub>6</sub> implies for the octahedral gaps between them an anti-phase rotation, which leaves this 'empty' octahedron nearly undistorted (Fig. 9a). These rotational modes do not affect the cations, therefore it is insignificant whether the octahedron is 'empty' or 'filled'. For the analogy with the perovskites it is important that all octahedra remain essentially undistorted, that is, we find in both structures an identical scheme of corner-linked tilted  $X_6$  octahedra. In consequence, the distribution of the diffuse X-ray intensity due to this type of motion is nearly identical in both cases. The streaks in cubic perovskites extend parallel to the principal axis from the R to the M point of the Brillouin zone of the primitive cubic lattice. The coordinates of the R and M points are  $(h + \frac{1}{2}, k + \frac{1}{2}, l + \frac{1}{2})$  $\frac{1}{2}$ ) and  $(h + \frac{1}{2}, k + \frac{1}{2}, l)$ . Referred to the reciprocal lattice of  $K_2$ SnCl<sub>6</sub> with  $a^*$  (perovskite) =  $2a^*$  ( $K_2$ SnCl<sub>6</sub>), these points are indexed (2h + 1, 2k + 1, 2l + 1) and (2h + 1, 2k + 1, l), respectively. So the R point in the perovskite cell corresponds to an odd-indexed reciprocal-lattice point in the f.c.c.  $R_2MX_6$  cell and the M point to the X point of the latter. The streaks of the perovskite structure from R to M correspond to those along  $\Delta$  from  $\Gamma$  to X between the odd-indexed reflections of the  $R_2MX_6$  cell.

The streaks along  $\Delta$  between the even-indexed reflections of the  $R_2MX_6$  structure have no analog in the perovskites, so they display one difference between the structures. The inelastic structure factor for those streaks is proportional to the deviation of the *u* parameter from 1/4 (see Tables 1 and 2). This means that the streaks are caused by the difference between the diameters of the *M*-filled  $MX_6$  and the empty  $X_6$ octahedra in the  $R_2MX_6$  structure.

As a further difference we expect that the rotational mode in perovskites is of higher energy than in  $R_2MX_6$ , because higher librational amplitudes expand the linked perovskite octahedra, that is, the rotation of the octahedra couples with an internal mode of the octahedra. The 'filled'  $MX_6$  octahedra in  $R_2MX_6$  rotate as rigid units until considerable steric hinderance sets in ( $\varphi \simeq$ 15°). The empty octahedra may expand, since the



Fig. 8. Electrostatic and hard-core lattice energy as a function of the rotation angle of the octahedra for an in-phase rotation of adjacent octahedra in a plane. Parameters are the Born-Meyer repulsive constants  $\rho_{12}$ ,  $\lambda_{12}$  for the K-Cl and Cl-Cl potentials.  $\Phi_{12} = -q_1 q_2/r_{12} + \lambda_{12} \exp(-r_{12}/\rho_{12}), q_1$  and  $q_2$  are the charges with distance vector  $r_{12}$ . Curve (1):  $\lambda_{\text{KCl}} = 3.63$ ,  $\rho_{\text{KCl}} = 0.324$  (taken from the KCl structure); the Cl-Cl repulsive parameters are arbitrary,  $\lambda_{\text{ClCl}} = 4.0$ ,  $\rho_{\text{ClCl}} = 0.363$ . Curve (2) shows that the minimum vanishes if the  $\rho_{\text{KCl}}$  parameter is raised to 0.37. The contribution of the Coulomb energy is shown in Fig. 1(*a*). Curves (3*a*) and (3) show the Coulomb and lattice energy, respectively, for the anti-phase rotation of adjacent octahedra in a plane. The repulsive parameters are those of curve (1).

Coulomb force is repelling the Cl ions and since no covalent forces are holding them together.

The second diffuse system in the  $\Sigma$  direction in  $K_2 \text{SnCl}_6 X$ -ray photographs is assumed to be typical for  $R_2 M X_6$  structures. The  $B_1(\Sigma)$  mode elongates adjacent  $M X_6$  octahedra normal to their distance vector and the 'empty' corner-linked  $X_6$  octahedra are distorted (Fig. 9b), so one would expect that in structures with all the octahedra filled this type of motion needs a higher excitation energy. On the other hand, we cannot expect to find in  $R_2 M X_6$  photographs the diffuse planes characteristic of the perovskite structure, because the correlated *M*-chain movement requires corner-linked  $M X_6$  octahedra.

To get a model for the phase transition in K<sub>2</sub>SnCl<sub>6</sub> at  $T_{c1} = 261$  K we assume that the potential with respect to the octahedra rotation exhibits two minima located symmetrically to the average position at  $\varphi = 0$ . The double-minimum potential is anharmonic; the difference-Fourier map confirms that the thermal ellipsoids of the harmonic approximation are insufficient to describe the librational amplitudes of the octahedra. At  $T > T_{c1}$ , the energy of the torsional mode is well above the potential barrier at  $\varphi = 0$ .

The broad potential well yields large amplitudes and the repulsion of the Cl ions by the K ions stabilizes the lattice. With decreasing temperature the lattice constant decreases. This implies an increase in the Cl parameter, u, when rigid octahedra are assumed. The latticeenergy calculations show that with increasing u and decreasing lattice constant a the potential well at  $\varphi = 0$ is raised, so we expect the first phase transition at  $T_{c1}$ when the mode energy is insufficient to penetrate the potential barrier.

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Fig. 9. Deformation of the 'empty' octahedra in  $K_2 \text{SnCl}_6$  between the rotated  $\text{SnCl}_6$  octahedra (a) for the  $A_2(\varDelta)$  mode  $[\mathbf{k} = (010)]$ and (b) for the  $B_1(\varSigma)$  mode  $[\mathbf{k} = (\frac{11}{22}0)]$ .

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## An Electron Diffraction Study of the Molecular Structure of the Sodium Chloride Dimer

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#### Abstract

The structure of sodium chloride has been studied by gas-phase electron diffraction using photographic plates designed for high-temperature work [Kakumoto, Ino, Kodera & Kakinoki (1977). J. Appl. Cryst. 10, 100–103]. The molecular structure at about 1130 K has been determined by an analysis based on the new complex scattering factors for Na<sup>+</sup> and Cl<sup>-</sup> calculated recently by the present authors. The radial distribution function shows the existence of a considerable amount of dimer molecules in the vapor. The structures of the monomer and dimer have been analyzed by a least-

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squares method assuming that the dimer is of a planar diamond shape: for monomer  $r_a(Na-Cl) = 2.392 \pm 0.028$ , for dimer  $r_a(Na-Cl) = 2.515 \pm 0.017$ ,  $r_a(Cl-Cl) = 3.893 \pm 0.021$  Å,  $\angle ClNaCl = 101.4 \pm 0.8^{\circ}$ . The present study indicates that the Na-Cl distance of the dimer is longer than that of the monomer and the determined structure parameters differ appreciably from existing theoretical predictions.

## Introduction

The structure of alkali halide molecules at high temperatures was studied by electron diffraction for the first time by Maxwell, Hendricks & Mosley (1937). They determined the internuclear distances in various

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